## Thermally oxidized Pd loaded WO<sub>3</sub> nanowires for Hydrogen detection

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## Abstract

Recently, environmental monitoring has become even more pressing need due to the air pollution caused by increasing number of automobiles and industry. In this regard, metal oxide semiconductor-based gas sensors have attracted a great deal of attention due to their high sensitivity, high stability, low cost and easy fabrication. High surface to volume ratio can significantly increase the absorption of the gas molecules and thus growth of 1-D nanostructures is important in enhancing the gas sensing performance. Therefore, WO<sub>3</sub> nanowires were synthesized on alumina substrate (2 mm  $\times$  2 mm) by thermal oxidation, which is simple, low cost and catalyst free technique that can be used to grow 1-D nanostructures. RF magnetron sputtering was used to deposit W layers (25 nm), and thermal oxidation was carried out at 600 °C in an oxygen deficient environment. Further, Pd loading was performed by RF magnetron sputtering at room temperature. XRD revealed that asprepared WO<sub>3</sub> nanowires were crystalline with a monoclinic phase of WO<sub>3</sub>. The structural evaluation showed that oxygen flow plays a crucial role in the growth of WO<sub>3</sub> nanowires. Furthermore, the oxidation temperature and oxidation time have a significant effect on the growth of WO<sub>3</sub>. Gas sensing measurements showed that  $Pd-WO_3$  nanowires have the ability to detect  $H_2$  at a low temperature range of 150 -250 °C. Optimum working temperatures for H<sub>2</sub> and H<sub>2</sub>S detection were identified as 250 °C and 350 °C, respectively, and a selective detection can be achieved by changing the temperature. These results suggest that Pd-WO<sub>3</sub> sensors are good candidate for detection of H<sub>2</sub> at low temperature range.

Keywords: WO<sub>3</sub> nanowires, Gas sensing, H<sub>2</sub> Sensing, Air pollution, H<sub>2</sub>S Sensing.